

This article was downloaded by:

On: 26 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

Selective Metalation of 6-Methylpurines: Synthesis of 6-Fluoromethylpurines and Related Nucleosides

Abdalla E. A. Hassan^a; William B. Parker^a; Paula W. Allan^a; John A. Montgomery^a; John A. Secrist III^a

^a Drug Discovery Division, Southern Research Institute, Birmingham, Alabama, USA

Online publication date: 09 August 2003

To cite this Article Hassan, Abdalla E. A. , Parker, William B. , Allan, Paula W. , Montgomery, John A. and Secrist III, John A. (2003) 'Selective Metalation of 6-Methylpurines: Synthesis of 6-Fluoromethylpurines and Related Nucleosides', *Nucleosides, Nucleotides and Nucleic Acids*, 22: 5, 747 – 749

To link to this Article: DOI: 10.1081/NCN-120022625

URL: <http://dx.doi.org/10.1081/NCN-120022625>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Selective Metalation of 6-Methylpurines: Synthesis of 6-Fluoromethylpurines and Related Nucleosides

Abdalla E. A. Hassan,* William B. Parker, Paula W. Allan,
John A. Montgomery, and John A. Secrist III

Drug Discovery Division, Southern Research Institute,
Birmingham, Alabama, USA

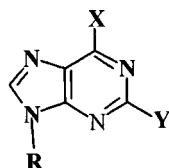
ABSTRACT

A selective metalation at the 6-CH₃ over C-8 of 6-methylpurine derivative **6** was observed with softer counter cation (Na⁺ or K⁺) of the base, while the harder Li⁺ showed no selectivity. In the presence of *N*-fluorobenzenesulfonamide (NFSI), this property was utilized for the synthesis of 6-fluoromethylpurine derivatives **4** and **5** as potential toxins for suicide gene therapy.

We have developed a cancer gene therapy strategy that is based on the activation of a non-toxic purine nucleoside (prodrug) to a highly toxic purine analog by a non-human gene, *E. coli* purine nucleoside phosphorylase (*E. coli* PNP), which is selectively expressed in tumor cells.^[1] *E. coli* PNP differs from human PNP in its ability to accept not only 6-oxopurine nucleosides, but also 6-aminopurine and certain adenine nucleoside analogs as substrates. This property has been used to cleave non-toxic adenine nucleoside analogs to very toxic adenine analogs, which would readily diffuse across cell membranes and have high bystander activity.^[1c] The toxic

*Correspondence: Abdalla E. A. Hassan, Pharmasset Inc., 1860 Montreal Road, Tucker, GA 30084, USA; Fax: +1 678 395 0039; E-mail: ahassan@services.pharmasset.com.





1: X = CH₃, Y = H, R = H

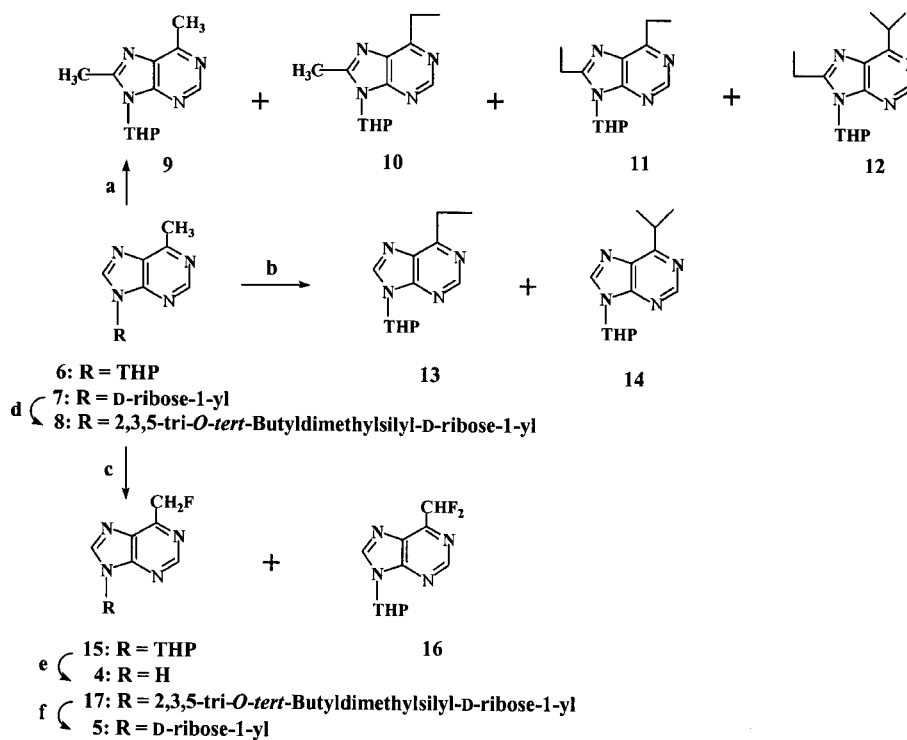
2: X = NH₂, Y = F, R = H

3: X = CH₃, Y = H, R = D-ribose-1-yl

4: X = CH₂F, Y = H, R = H

5: X = CH₂F, Y = H, R = D-ribose-1-yl

Chart 1.



Scheme 1. ^aReagents and conditions: a) Base (*n*-BuLi, or LDA, or LTMP, or LiHMDS), THF, -78°C, excess CH₃I; b) Base (NaHMDS or KHMDS), THF, -78°C, excess CH₃I; c) NaHMDS, THF, -78°C, 30 min, then NFSI, 48–56%; d) *tert*-Butyldimethylsilyl chloride, Im, DMF, r. t., 84%; e) 1N HCl, THF, r. t., 94%; f) Et₄NF, CH₃CN, r. t., 95%.

adenine analogs of most interest to date are 6-methylpurine (6-MeP, **1**) and 2-fluoro-adenine (**2**) (Chart 1); however, we still continue to search for the optimal toxin/prodrug combination that would have the desired biological properties. Herein, we report on the selective metalation at the 6-CH₃ moiety of 6-methylpurine derivatives and the utilization of this property for the synthesis of 6-fluoromethylpurine (6-FMeP, **4**) and related nucleosides of potential use for this project.

Lithiation⁵ of **6**^[1d] with *n*-BuLi in THF at -78°C, in the presence of MeI resulted in the formation of a mixture of 6,8-dimethylpurine derivative, **9** as a major product along with a mixture of compounds **10**, **11**, and **12**, respectively. A non-selective lithiation was also observed with LDA, LiHMDS, or LTMP in THF at -78°C, irrespective of the molar equivalence of the base. On the other hand, when the base was changed to NaH, or (CH₃)₃COK at 0°C in the presence of MeI, a mixture of 6-ethylpurine derivative **13** (major) and 6-isopropyl derivative **13** (minor) was obtained in good yield. A similar selectivity at the 6-CH₃ position was also observed with NaHMDS or KHMDS at -78°C in THF (Sch. 1). Quenching the sodium salt of **6** (generated by NaHMDS at -78°C in THF) with NFSI gave the 6-FMeP derivative **15** in good yield along with traces of **16** (Sch. 1). The 6-FMe-P riboside derivative **17** was also synthesized by applying the same chemistry on compound **8** (Sch. 1). Deprotection of **15** and **17** under conventional conditions gave **4** and **5** in good yields, respectively.

The newly synthesized compounds were evaluated for their cytotoxic activity as well as their substrate activity to *E. coli* PNP. 6-FMe-P (**4**) showed potent cytotoxic activity against CCRF-CEM cells (2 μM) and moderate activity against our solid tumor panel. Furthermore, the riboside derivative **5** showed potent cytotoxic activity against CCRF-CEM cells (0.03 μM) and was also as good a substrate as the parent compound **2** in terms of substrate activity to *E. coli* PNP.

REFERENCES

1. a) Sorscher, E.J.; Peng, S.; Bebok, Z.; Allan, P.W.; Bennett, L.L., Jr.; Parker, W.B. *Gene Therapy* **1994**, *1*, 233–238; b) Parker, W.B.; King, S.A.; Allan, P.W.; Bennett, L.L., Jr.; Secrist III, J.A.; Montgomery, J.A.; Gilbert, K.S.; Waud, W.R.; Wells, A.H.; Gillespie, G.Y.; Sorscher, E.J. *Human Gene Therapy*, **1997**, *8*, 1637–1644; c) Hughes, B.W.; King, S.A.; Allan, P.W.; Parker, W.B.; Sorscher, E.J. *J. Biol. Chem.* **1998**, *273*, 2322–2328; d) Hassan, A.E.A.; Abou-Elkhair, R.A.I.; Montgomery, J.A.; Secrist III, J.A. *Nucleosides, Nucleotides & Nucleic Acids* **2000**, *19* (7), 1123–1134.
2. Komamoto, H.; Tanaka, H.; Takioka, R.; Ishida, Y.; Nakamura, A.; Kimura, S.; Hayakawa, H.; Kato, K.; Miyasaka, T. *J. Org. Chem.* **1999**, *64*, 7773–7780.



